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Original Article

Modelling the Influence of Evaporation on Residual Chlorine in Water Storage Tanks Using CFD.

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Water storage tanks are usually utilized in water distribution systems (WDS) 08 July 2021 to meet the water demand fluctuations. Chlorine is the most common disinfectant used to disinfect water supplies. However, variations in the rate Keywords: of chlorine decay in these storage tanks are one of the greatest limiting factors in ensuring adequate water treatment process and giving guarantee to its Chlorine Decay, efficiency. These variations could be due to some inadequately tested Water Treatment. mechanisms of chlorine reactions in bulk fluid, chlorine reactions with CFD, storage tank walls, and natural evaporation. This study presents OpenFoam, Computational Fluid Dynamics (CFD) modelling approach to assess the Evaporation. influence of evaporation on residual chlorine in water storage tanks. Findings indicate that an increase in the evaporation rate accelerates the rate at which residual chlorine is lost. It is concluded that temperature is the main factor influencing evaporation, which in turn causes the disappearance of residual chlorine within the water storage tanks.

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| α | Phase fraction | |
|-------------|--|---------------------|
| ρ | Density [kgm ⁻³] | |
| κ | Curvature | |
| k | Thermal conductivity [<i>Wm</i> ⁻¹] | K^{-1}] |
| μ; | Fluid viscosity $[m^2s^{-1}]$ | |
| m`e | Rate of evaporation $[s^{-1}]$ | |
| η; | First-order decay constant | $[s^{-1}]$ |
| $\eta_{b;}$ | bulk decay rate constant | $[s^{-1}]$ |
| $\eta_{w;}$ | wall decay rate constant | $[s^{-1}]$ |
| $C_{p;}$ | Specific heat capacity | $[Jmol^{-1}K^{-1}]$ |

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